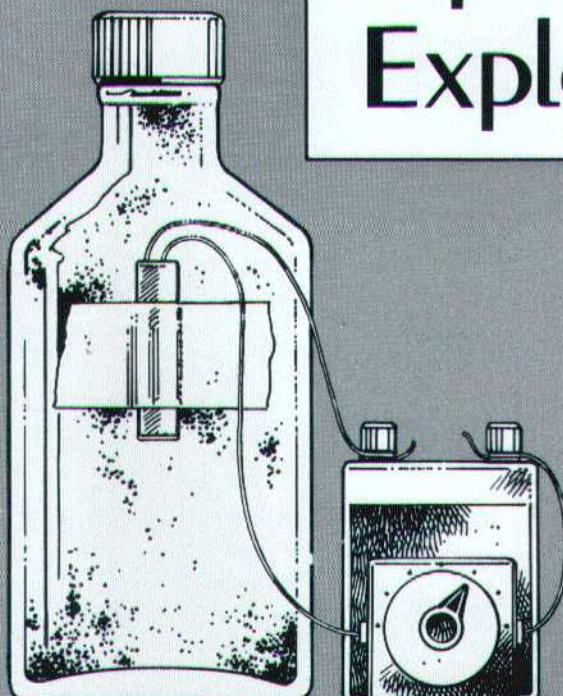


# INCENDIARIES

Advanced  
Improvised  
Explosives



SEYMOUR LECKER

**FIRE HAS A LONG HISTORY OF USE BY TERRORISTS AND SABOTEURS BECAUSE OF ITS POTENTIAL FOR MAJOR DESTRUCTION FROM MINOR EFFORT. POUND FOR POUND, INCENDIARIES CAN DO MORE DAMAGE THAN EXPLOSIVES, IF PROPERLY USED. ARSON HAS ALSO PROVEN ITSELF AS AN EFFECTIVE PSYCHOLOGICAL WEAPON, PLAYING UPON MAN'S ANCIENT AWE AND FEAR OF FIRE. THE TIME LAG BETWEEN THE START OF THE FIRE AND THE DESTRUCTION OF THE TARGET CAN BE AN ADVANTAGE IN CERTAIN OPERATIONS.**

**IN THIS THIRD VOLUME OF THE ADVANCED IMPROVISED EXPLOSIVES SERIES, AUTHOR SEYMOUR LECKER DETAILS THE USE OF WIDELY USED, RELATIVELY INEXPENSIVE INDUSTRIAL CHEMICALS THAT BURN WITH INTENSE HEAT. ALSO INCLUDED IS INFORMATION ON OXIDIZERS AND IGNITION DEVICES, SAFETY AND HEALTH HAZARDS, AND A USEFUL GLOSSARY OF CHEMICAL TERMS.**

**WARNING: INCENDIARY DEVICES HAVE THE POTENTIAL TO BE MORE DANGEROUS FOR THE MAKER THAN THE INTENDED TARGET, AND THE USE OF PROPER SAFEGUARDS AND RESTRAINTS IS ESSENTIAL.  
THIS MANUAL IS PRESENTED FOR INFORMATION PURPOSES ONLY.**

A Paladin Press Book  
ISBN 0-87364-483-2

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# SEYMOUR LECKER

PALADIN PRESS  
BOULDER, COLORADO

**Also by Seymour Lecker:**

Deadly Brew: Advanced Improvised Explosives

Improvised Explosives: How to Make Your Own

Shock Sensitive Industrial Materials: Advanced  
Improvised Explosives

*Incendiaries:*

*Advanced Improvised Explosives*  
by Seymour Lecker

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ISBN 0-87364-483-2

Printed in the United States of America

Published by Paladin Press, a division of  
Paladin Enterprises, Inc., P.O. Box 1307,  
Boulder, Colorado 80306, USA.  
(303) 443-7250

Direct inquiries and/or orders to the above address.

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Illustrations by Bill Border

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# **Warning**

This manual is for informational purposes only. Neither the author nor the publisher assumes any responsibility for the use or misuse of information in this book.

"Whoever maliciously damages or destroys or attempts to damage or destroy by means of an explosive or fire any real or personal property...

- Shall be imprisoned for not more than 10 years or fined not more than \$10,000 or both.
- If personal injury results, shall be imprisoned for not more than 20 years or fined not more than \$20,000 or both.
- If death results, shall be imprisoned for life or shall be subject to the death penalty."

— Federal Law Relating to Explosives

# Preface

This volume demonstrates the potential of commonly used industrial chemicals as incendiaries.

Fifty chemicals have been selected having a wide variety of industrial applications, but there are many more. These fifty were selected because they are widely used and have a flash point of under 100° Fahrenheit.

Fire has a long history of use by terrorists and saboteurs because of the potential for major destruction from minor effort. Arson has also proven itself as an effective psychological weapon playing on man's ancient awe and fear of fire.

Once again, it is impossible to overemphasize the potential for danger to persons attempting to use hazardous materials without proper safeguards and restraints.

## CHAPTER 1

---

# Industrial Chemicals

### ACETALDEHYDE $\text{CH}_3\text{CHO}$

#### **Synonyms**

Acetic Aldehyde, Ethanol, Ethanal, Ethyl Aldehyde

#### **Description**

Colorless fuming liquid, pungent fruitlike odor.

#### **Uses**

Manufacture of acetic acid and numerous other chemicals. Synthetic flavors.

#### **Hazards**

Moderately toxic (narcotic). Will cause irritation of eyes, skin, and respiratory tract.

#### **Flash point**

-36° F

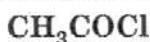
#### **Fire fighting**

$\text{CO}_2$ , dry chemical, alcohol foam.

#### **Ignition**

Flame or contact with acid anhydrides, alcohols, ketones, phenols, halogens, isocyanates, or strong alkalis.

## ACETYL CHLORIDE



### *Synonym*

Ethanoyl Chloride

### *Description*

Colorless, fuming liquid. Strong odor.

### *Uses*

Manufacture of dyestuffs and pharmaceuticals.

### *Hazards*

Highly toxic. Can irritate skin, eyes, and mucous membrane. Heat can cause toxic fumes.

### *Flash point*

40° F

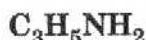
### *Fire fighting*

CO<sub>2</sub> or dry chemical.

### *Ignition*

Flame or contact with water or ethanol.

## ALLYL AMINE



### *Synonym*

2-Propenylamine

### *Description*

Colorless to light yellow liquid. Ammonia-like odor.

### *Uses*

Manufacture of pharmaceuticals. Organic synthesis.

### *Hazards*

Highly toxic. Can irritate eyes, skin, and respiratory system. Heat can cause toxic fumes.

### *Flash point*

-20° F

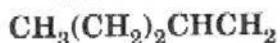
***Fire fighting***

Alcohol foam, CO<sub>2</sub>, dry chemical.

***Ignition***

Flame.

**AMYLENE**



***Synonyms***

Propylethylene, 2-Methyl Butene-2, 1-Pentene

***Description***

Colorless liquid, highly disagreeable odor.

***Uses***

Organic synthesis, blending agent for high-octane motor fuels, manufacture of pesticides.

***Hazards***

Moderately toxic. Narcotic in high concentrations.

Can cause asphyxiation. Heat can cause toxic fumes.

***Flash point***

0° F

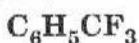
***Fire fighting***

Alcohol foam, water spray or mist, dry chemical.

***Ignition***

Flame or contact with powerful oxidizers.

**BENZOTRIFLUORIDE**



***Synonyms***

Toluene Trifluoride, Trifluoromethylbenzene

***Description***

Water-white liquid. Aromatic odor.

### ***Uses***

Intermediate for dyes and pharmaceuticals, solvent, dielectric fluid, vulcanizing agent, and insecticide.

### ***Hazards***

Highly toxic. Can irritate skin, eyes, and mucous membrane.

### ***Flash point***

54° F

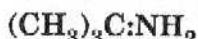
### ***Fire fighting***

Foam, CO<sub>2</sub>, water spray, mist, dry chemical.

### ***Ignition***

Flame and contact with oxidizing materials.

## **TERT - BUTYLAMINE**



### ***Synonyms***

2-Aminoisobutane, Trimethyl Aminomethane

### ***Description***

Colorless liquid.

### ***Uses***

Manufacture of insecticides, fungicides, dyestuffs, and pharmaceuticals. Intermediate for rubber accelerators.

### ***Hazards***

Highly toxic. Will irritate skin on contact.

### ***Flash point***

50° F

### ***Fire fighting***

Alcohol foam.

### ***Ignition***

Flame.

**TERT - BUTYL HYDROPEROXIDE**  
 $(\text{CH}_3)_3\text{COOH}$

**Synonyms**

None

**Description**

Water-white liquid.

**Uses**

Polymerization, oxidation, sulfanation catalyst, bleaching, and deodorizing.

**Hazards**

Highly toxic. Can cause severe depression and incoordination. High concentrations can cause respiratory arrest.

**Flash point**

80° F

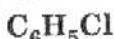
**Fire fighting**

Alcohol foam,  $\text{CO}_2$ , dry chemical.

**Ignition**

Flame or contact with reducing materials.

**CHLOROBENZENE**



**Synonyms**

Chlorobenzol, Monochlorobenzene, Phenyl Chloride

**Description**

Colorless liquid. Almond-like odor.

**Uses**

Solvent, pesticide intermediate, heat transfer, manufacture of phenol, chloronitrobenzene, and aniline.

**Hazards**

Moderately toxic. Narcotic in high concentrations.

***Flash point***

85° F

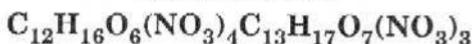
***Fire fighting***

CO<sub>2</sub>, dry chemical, water, foam.

***Ignition***

Flame or contact with oxidizing materials.

**COLLODION**



***Synonyms***

None

***Description***

Pale yellow, syrupy liquid, ethereal odor.

***Uses***

Cements, coating wounds, solvent for drugs, corn removers, process engraving, lithography, photography.

***Hazards***

Material is relatively safe, but moderate heat can create large quantities of extremely toxic gases.

***Flash point***

0° F

***Fire fighting***

Alcohol foam.

***Ignition***

Flame.

## CYCLOHEXANE



### *Synonyms*

Hexamethylene, Hexanaphthene, Hexahydrobenzene

### *Description*

Colorless liquid. Pungent odor.

### *Uses*

Manufacture of nylon, solvent, extracting essential oils, organic synthesis, recrystallization, paint and varnish remover, fungicide.

### *Hazards*

Moderately toxic. Can irritate skin. Narcotic in high concentrations.

### *Flash point*

-4° F

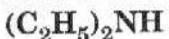
### *Fire fighting*

Foam, CO<sub>2</sub>, dry chemical, spray, fog.

### *Ignition*

Flame or contact with powerful oxidizers.

## DIETHYLAMINE



### *Synonyms*

None

### *Description*

Colorless liquid, ammonia-like odor.

### *Uses*

Manufacture of rubber chemicals, textile specialties, solvents, flotation agents, resins, pesticides, polymerization inhibitors, dyes, pharmaceuticals, petroleum chemicals, and corrosion inhibitors. Electroplating.

### ***Hazards***

Moderately toxic. Can irritate skin and mucous membrane.

### ***Flash point***

-9° F

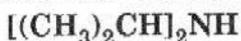
### ***Fire fighting***

Alcohol foam, CO<sub>2</sub>, dry chemical.

### ***Ignition***

Flame or contact with oxidizing materials.

## **DIISOPROPYLAMINE**



### ***Synonyms***

None

### ***Description***

Colorless liquid. Amine-like odor.

### ***Uses***

Chemical intermediate catalyst.

### ***Hazards***

Moderately toxic. Can irritate skin, eyes, and respiratory tract.

### ***Flash point***

30° F

### ***Fire fighting***

Alcohol foam, foam, CO<sub>2</sub>, dry chemical.

### ***Ignition***

Flame or contact with oxidizing materials.

**DIMETHYL SULFIDE**  
**(CH<sub>3</sub>)<sub>2</sub>S**

***Synonyms***

Methyl Sulfide, Methyl Thiomethane

***Description***

Colorless to straw-colored liquid. Disagreeable odor.

***Uses***

Gas odorant, solvent for many inorganic substances, catalyst impregnator.

***Hazards***

Moderately toxic. Will irritate eyes. Heat can cause highly toxic fumes.

***Flash point***

0° F

***Fire fighting***

Dry chemical, foam, CO<sub>2</sub>.

***Ignition***

Flame and contact with powerful oxidizers.

**DIOXOLANE**  
**C<sub>3</sub>H<sub>6</sub>O<sub>2</sub>**

***Synonym***

Ethylene Glycol Formal

***Description***

Water-white liquid.

***Uses***

Low-boiling solvent and extractant for oils, fats, waxes, dyes, and cellulose derivatives.

***Hazards***

Moderately toxic. Can irritate skin, eyes, and respiratory tract.

***Flash point***

35° F

***Fire fighting***

Alcohol foam, CO<sub>2</sub>, dry chemical.

***Ignition***

Flame or contact with powerful oxidizers.

**ETHYLAMINE**



***Synonyms***

Aminoethane, Monoethylamine

***Description***

Colorless liquid. Strong ammonia odor.

***Uses***

Dye intermediate, solvent extraction, refining petroleum, stabilizing latex, detergents, organic synthesis.

***Hazards***

Highly toxic. Will irritate or burn skin, eyes, and respiratory tract.

***Flash point***

0° F

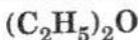
***Fire fighting***

Alcohol foam, dry chemical.

***Ignition***

Flame or contact with oxidizing materials.

## **ETHYL ETHER**



### ***Synonyms***

Diethyl Ether, Diethyl Oxide, Ether, Ethyl Oxide,  
Sulfuric Ether

### ***Description***

Colorless liquid. Sweet aromatic odor.

### ***Uses***

Organic synthesis, analytical chemistry, anesthetic,  
extractant, manufacture of smokeless powder and  
industrial solvents.

### ***Hazards***

Moderately toxic. Can cause drowsiness, depression,  
and unconsciousness.

### ***Flash point***

-49° F

### ***Fire fighting***

Alcohol foam, CO<sub>2</sub>, dry chemical.

### ***Ignition***

Flame, static electricity, or strong oxidizing  
materials.

## **ETHYL MERCAPTAN**



### ***Synonyms***

Ethanethiol, Ethylhydrosulfide, Ethyl Thioalcohol,  
Ethylsulphydrate

### ***Description***

Colorless liquid. Odor of skunk.

### ***Uses***

Odorant, adhesive, stabilizer, chemical intermediate.

**Hazards**

Moderately toxic. Odor tends to cling. Heat or contact with water or acid will cause highly toxic fumes.

**Flash point**

80° F

**Fire fighting**

CO<sub>2</sub>, dry chemical.

**Ignition**

Flame, spark, or contact with oxidizing materials.

**ETHYL NITRATE**



**Synonym**

Nitric Ether

**Description**

Colorless liquid, pleasant sweet odor.

**Uses**

Manufacture of drugs, perfumes, and dyes. Organic synthesis, rocket propellant.

**Hazards**

Moderately toxic. Can cause dizziness, abdominal cramps, diarrhea, and depression.

**Flash point**

50° F

**Fire fighting**

Foam, CO<sub>2</sub>, dry chemical.

**Ignition**

Flame.

## ETHYL NITRITE



### *Synonyms*

Nitrous Ether, Hyponitrous Ether

### *Description*

Colorless or yellowish liquid, ethereal odor.

### *Uses*

Organic reactions, synthetic flavoring.

### *Hazards*

Moderately toxic. Can cause increased pulse rate, decreased blood pressure, and unconsciousness.

### *Flash point*

-31° F

### *Fire fighting*

Foam, CO<sub>2</sub>, dry chemical.

### *Ignition*

Flame. Intense light can cause detonation.

## FLUOROBENZENE



### *Synonym*

Phenyl Fluoride

### *Description*

Colorless liquid. Odor of benzene.

### *Uses*

Insecticide and larvicide intermediate, identification reagent for plastic and resin polymers.

### *Hazards*

Highly toxic. Will irritate skin, eyes, and mucous membrane.

***Flash point***

5° F

***Fire fighting***

Water spray, mist, foam, dry chemical, CO<sub>2</sub>.

***Ignition***

Flame or contact with oxidizing materials.

**FURAN**

C<sub>4</sub>H<sub>7</sub>O

***Synonyms***

Furfuran, Fur furane, Oxole, Tetrol

***Description***

Colorless to reddish-brown, oily liquid. Almond-like odor.

***Uses***

Organic synthesis.

***Hazards***

Moderately toxic. Narcotic. Can cause eye and throat irritation as well as headaches.

***Flash point***

32° F

***Fire fighting***

CO<sub>2</sub>, dry chemical.

***Ignition***

Flame or contact with acids or oxidizing materials.

## FURFURLYLAMINE



### *Synonym*

2-Furanmethylamine

### *Description*

Straw-colored liquid. Faint odor of amine.

### *Uses*

Corrosion inhibitor, soldering flux, chemical intermediate.

### *Hazards*

Moderately toxic. Will irritate skin, eyes, and mucous membrane.

### *Flash point*

99° F

### *Fire fighting*

Foam, CO<sub>2</sub>, dry chemical.

### *Ignition*

Flame.

## 2-HEPTENE



### *Synonym*

2-Heptylene

### *Description*

Colorless liquid.

### *Uses*

Plant growth retardant.

### *Hazards*

Mildly toxic. Can be an irritant and/or asphyxiant.

### *Flash point*

32° F

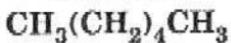
***Fire fighting***

Dry chemical, CO<sub>2</sub>, foam.

***Ignition***

Flame or contact with oxidizing materials.

**HEXANE**



***Synonym***

Hexyl Hydride

***Description***

Colorless liquid. Faint, pleasant odor.

***Uses***

Solvent for vegetable oils, thermometers (low temperature), polymerization, paint dilutor, and used to denature alcohol.

***Hazards***

Mildly toxic. Can cause vertigo, drowsiness, loss of appetite, blurred vision, and itching.

***Flash point***

-7° F

***Fire fighting***

CO<sub>2</sub>, dry chemical.

***Ignition***

Flame and contact with oxidizing materials.

## HYDROCYANIC ACID HCN

### ***Synonyms***

Prussic Acid, Hydrogen Cyanide, Formonitrile

### ***Description***

Colorless to white liquid. Faint odor of bitter almonds.

### ***Uses***

Manufacture of acrylonitrile, acrylates, adiponitrile, cyanide salts, dyes, chelates, rodenticides, and pesticides.

### ***Hazards***

Extremely toxic. Limited exposure will cause dizziness, headaches, and nausea. Continued exposure will quickly result in death.

### ***Flash point***

0° F

### ***Fire fighting***

CO<sub>2</sub>, non-alkaline dry chemical, foam.

### ***Ignition***

Flame or contact with acetaldehyde.

## ISOPENTANE $(\text{CH}_3)_2\text{CHCH}_2\text{CH}_3$

### ***Synonyms***

2-Methylbutane, Ethyldimethylmethane, Isomyl Hydride

### ***Description***

Colorless liquid, pleasant odor.

***Uses***

Solvent, blowing agent for polystyrene, manufacture of chlorinated derivatives.

***Hazards***

Moderately toxic. Can cause irritation, itching, blisters, and swelling.

***Flash point***

-60° F

***Fire fighting***

Foam, CO<sub>2</sub>, dry chemical.

***Ignition***

Flame or contact with oxidizing materials.

## ISOPROPYLAMINE (CH<sub>3</sub>)<sub>2</sub>CHNH<sub>2</sub>

***Synonym***

2-Aminopropane

***Description***

Colorless liquid, amino odor.

***Uses***

Solvent, intermediate in synthesis of rubber accelerators, de-hairing agent, manufacture of pharmaceuticals, dyes, insecticides, bactericides, and textiles.

***Hazards***

Moderately toxic. Will cause irritation. Narcotic in high concentrations.

***Flash point***

-35° F

***Fire fighting***

Alcohol foam, foam, CO<sub>2</sub>, dry chemical.

***Ignition***

Flame or contact with oxidizing materials.

## ISOPROPYL CHLORIDE

$(\text{CH}_3)_2\text{CHCl}$

### *Synonym*

2-Chloropropane

### *Description*

Colorless, odorless liquid.

### *Uses*

Solvent, intermediate.

### *Hazards*

Mildly toxic. Can be used as a surgical anesthetic.

### *Flash point*

-26° F

### *Fire fighting*

Dry chemical,  $\text{CO}_2$ , mist.

### *Ignition*

Flame or contact with oxidizing materials.

## METHALLYL CHLORIDE

$\text{C}_4\text{H}_7\text{Cl}$

### *Synonyms*

Beta-Methylallyl Chloride, Methyl Allyl Chloride,  
2-Chlorobutene-2

### *Description*

Colorless to straw-colored liquid. Very sharp odor.

### *Uses*

Manufacture of insecticides, plastics, and pharmaceuticals. Fumigant for grains, tobacco, and soil.

### ***Hazards***

Moderately toxic. Can irritate eyes and skin. Heat will produce highly toxic fumes.

### ***Flash point***

11° F

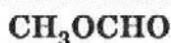
### ***Fire fighting***

Alcohol foam, CO<sub>2</sub>, dry chemical.

### ***Ignition***

Flame or contact with oxidizing materials.

## **METHYL FORMATE**



### ***Synonym***

Methyl Methanoate

### ***Description***

Colorless liquid. Agreeable odor.

### ***Uses***

Organic synthesis, cellulose acetate solvent, fumigant, larvicides, food additive.

### ***Hazards***

Moderately toxic. Can irritate eyes. Heat can cause toxic gases.

### ***Flash point***

-2° F

### ***Fire fighting***

Alcohol foam, CO<sub>2</sub>, dry chemical.

### ***Ignition***

Flame.

## **2-METHYLFURAN**



### ***Synonym***

Sylvan

### ***Description***

Colorless liquid. Ether-like odor.

### ***Uses***

Chemical intermediate.

### ***Hazards***

Highly toxic. Can irritate eyes, skin, and mucous membrane.

### ***Flash point***

-22° F

### ***Fire fighting***

CO<sub>2</sub>, dry chemical.

### ***Ignition***

Flame or contact with oxidizing materials.

## **NICKEL CARBONYL**



### ***Synonym***

Nickel Tetracarbonyl

### ***Description***

Colorless liquid or needles.

### ***Uses***

Production of high-purity nickel powder. Nickel coating of metals.

### ***Hazards***

Highly toxic. Can cause headache, dizziness, vomiting, fever, and breathing difficulties. Heat or contact with acids can cause toxic fumes.

### ***Flash point***

-4° F

### ***Fire fighting***

Water, foam, CO<sub>2</sub>, dry chemical.

### ***Ignition***

Flame or contact with oxidizing materials.

## **NONANE**



### ***Synonym***

Nonyl Hydride

### ***Description***

Colorless liquid.

### ***Uses***

Organic synthesis, biodegradable detergents, distillation chaser.

### ***Hazards***

Mildly toxic. Can irritate respiratory tract. Narcotic in high concentration.

### ***Flash point***

88° F

### ***Fire fighting***

CO<sub>2</sub>, dry chemical.

### ***Ignition***

Flame or contact with oxidizing materials.

**OCTANE**  
 $\text{CH}_3(\text{CH}_2)_6\text{CH}_3$

**Synonyms**

None

**Description**

Colorless liquid.

**Uses**

Solvent, organic synthesis, calibrations, and azeotropic distillation.

**Hazards**

Mildly toxic. Can asphyxiate. Narcotic in high concentration.

**Flash point**

56° F

**Fire fighting**

Alcohol foam,  $\text{CO}_2$ , dry chemical.

**Ignition**

Flame or contact with oxidizing materials.

**PROPANAL**  
 $\text{C}_2\text{H}_5\text{CHO}$

**Synonyms**

Propionaldehyde, Propyl Aldehyde, Propionic Aldehyde

**Description**

Colorless to white liquid. Fruity, suffocating odor.

**Uses**

Manufacture of plastics, synthetic rubber, and prionic acid. Disinfectant, preservatives.

***Hazards***

Moderately toxic. Will irritate skin, eyes, and mucous membranes.

***Flash point***

-22° F

***Fire fighting***

Alcohol foam, CO<sub>2</sub>, dry chemical.

***Ignition***

Flame or contact with oxidizing materials.

**PROPIONITRILE*****Synonyms***

Propionic Nitrile, Propanenitrile, Ethyl Cyanide

***Description***

Colorless liquid. Ethereal odor.

***Uses***

Solvent, intermediate and dielectric fluid.

***Hazards***

Highly toxic. Can cause headaches, visual distortions, fall in blood pressure, vomiting, collapse, and coma.

Heat will cause even more toxic fumes.

***Flash point***

36° F

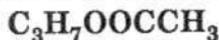
***Fire fighting***

Water spray, foam, mist, CO<sub>2</sub>, or dry chemical.

***Ignition***

Flame or contact with oxidizing materials.

## **PROPYL ACETATE**



### ***Synonym***

Propyl Ester

### ***Description***

Colorless liquid. Pleasant odor.

### ***Uses***

Manufacture of perfumes, flavoring agents, natural and synthetic resins, lacquers, and plastics. Organic synthesis, laboratory reagent, solvent for cellulose derivatives.

### ***Hazards***

Mildly toxic. Can cause irritations and retarded breathing.

### ***Flash point***

55° F

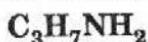
### ***Fire fighting***

Alcohol foam, CO<sub>2</sub>, dry chemical.

### ***Ignition***

Flame or contact with oxidizing materials.

## **PROPYLAMINE**



### ***Synonym***

1-Amino Propane

### ***Description***

Colorless liquid. Strong odor of ammonia.

### ***Uses***

Chemical intermediate, laboratory reagent.

### ***Hazards***

Mildly toxic. Can irritate or sensitize skin.

**Flash point**

-35° F

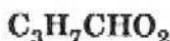
**Fire fighting**

Alcohol foam, CO<sub>2</sub>, dry chemical.

**Ignition**

Flame or contact with oxidizing materials.

**PROPYL FORMATE**



**Synonym**

Propyl Methanoate

**Description**

Colorless liquid. Pleasant odor.

**Uses**

Food flavoring.

**Hazards**

Mildly toxic. Can irritate skin and eyes.

**Flash point**

27° F

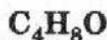
**Fire fighting**

Alcohol foam.

**Ignition**

Flame or contact with oxidizing materials.

**TETRAHYDROFURAN**



**Synonyms**

THF, Diethylene Oxide, Tetramethylene Oxide,  
Cyclotetramethylene Oxide-1, 4 Epoxy Butane

***Description***

Colorless to white liquid. Ether-like odor.

***Uses***

Manufacture of resins, vinyls, topcoating solutions, protective coatings, cellophane, adhesives, magnetic tapes, and printing ink. Chemical intermediate and monomer.

***Hazards***

Moderately toxic. Can irritate eyes and mucous membrane. High concentrations can be narcotic.

***Flash point***

6° F

***Fire fighting***

Foam, dry chemical, CO<sub>2</sub>.

***Ignition***

Flame or contact with oxidizers.

***Synonym***

Thiofuran

***Description***

Colorless liquid.

***Uses***

Manufacture of dyes and pharmaceuticals, organic synthesis, solvent.

***Hazards***

Moderately toxic. Will irritate eyes, skin, mucous membrane, and respiratory system.

***Flash point***

30° F

***Fire fighting***

Foam, CO<sub>2</sub>, dry chemical.

***Ignition***

Flame or contact with nitric acid.

**TRICHLOROSILANE**



***Synonym***

Silicochloroform

***Description***

Colorless liquid.

***Uses***

Purification of silicone.

***Hazards***

Mildly toxic. Heat can cause toxic fumes. Contact with water or steam can cause toxic and corrosive fumes.

***Flash point***

7° F

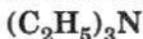
***Fire fighting***

CO<sub>2</sub>, dry chemical.

***Ignition***

Flame or contact with oxidizing materials.

## TRIETHYLAMINE



### *Synonyms*

None

### *Description*

Colorless liquid. Odor of ammonia.

### *Uses*

Corrosion inhibitor, propellant, catalytic solvent in chemical synthesis, accelerator activators for rubber, curing and hardening of polymers.

### *Hazards*

Highly toxic. Will irritate all tissue. Can damage kidney and liver.

### *Flash point*

16° F

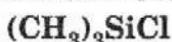
### *Fire fighting*

CO<sub>2</sub>, dry chemical, alcohol foam.

### *Ignition*

Flame or contact with powerful oxidizing materials.

## TRIMETHYL CHLOROSILANE



### *Synonyms*

None

### *Description*

Colorless liquid.

### *Uses*

Intermediate for silicone fluids.

### *Hazards*

Moderately toxic. Will irritate tissue.

***Flash point***

-18° F

***Fire fighting***

Foam, alcohol foam or fog.

***Ignition***

Flame or contact with water.

**VINYL BUTYL ETHER**



***Synonyms***

Butyl Vinyl Ether, Vinyl-N-Butyl Ether, N-Butyl  
Vinyl Ether

***Description***

Colorless liquid.

***Uses***

Copolymerization, synthesis.

***Hazards***

Mildly toxic. Can cause loss of appetite, thirst, and fatigue.

***Flash point***

15° F

***Fire fighting***

Foam, CO<sub>2</sub>, dry chemical.

***Ignition***

Flame or contact with powerful oxidizers.

## VINYLCYCLOHEXENE



### *Synonyms*

1-Vinylcyclohexene-3, 4-Vinylcyclohexene-1,  
Cyclohexenylethylene

### *Description*

Colorless liquid.

### *Uses*

Polymers, organic synthesis.

### *Hazards*

Mildly toxic. Can irritate tissue. Narcotic in high concentrations.

### *Flash point*

61° F

### *Fire fighting*

Foam, CO<sub>2</sub>, dry chemical.

### *Ignition*

Flame or contact with oxidizing materials.

## VINYL ISOBUTYL ETHER



### *Synonyms*

Isobutyl Vinyl Ether, Ive

### *Description*

Colorless liquid.

### *Uses*

Manufacture of surgical adhesives, lacquers, resins, and plastics. Chemical intermediate.

### *Hazards*

Mildly toxic. Can cause loss of appetite, thirst, and fatigue.

***Flash point***

15° F

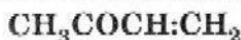
***Fire fighting***

Alcohol foam, CO<sub>2</sub>, dry chemical.

***Ignition***

Flame or contact with oxidizing materials.

**VINYL METHYL KETONE**



***Synonyms***

3-Butene-2-One, Methyl Vinyl Ketone

***Description***

Colorless liquid. Irritating odor.

***Uses***

Synthesis and alkylating agent, intermediate in the manufacture of steroids and vitamins, monomer for vinyl resins.

***Hazards***

Moderately toxic. Can irritate skin, eyes, and mucous membrane. Heat can create toxic fumes.

***Flash point***

20° F

***Fire fighting***

CO<sub>2</sub>, dry chemical.

***Ignition***

Flame or contact with oxidizing materials.

## VINYL PROPIONATE



### *Synonyms*

None

### *Description*

Colorless liquid.

### *Uses*

Manufacture of emulsion paints.

### *Hazards*

Mildly toxic. Can cause loss of appetite, thirst, and fatigue.

### *Flash point*

34° F

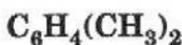
### *Fire fighting*

Alcohol foam.

### *Ignition*

Flame or contact with oxidizing materials.

## XYLENE



### *Synonyms*

Dimethyl Benzene, Xylool

### *Description*

Colorless liquid.

### *Uses*

Solvent, insecticides, synthesis of organic chemicals. Manufacture of protective coatings, enamels, rubber cements, dyes, motor fuels, pharmaceuticals, and vitamins.

***Hazards***

Moderately toxic. Can irritate skin, eyes, and mucous membrane.

***Flash point***

81° F

***Fire fighting***

Foam, CO<sub>2</sub>, dry chemical.

***Ignition***

Flame or contact with powerful oxidizers.

CHAPTER 2



## Oxidizers

Many of the materials outlined in Chapter 1 will ignite upon contact with oxidizers. Many of the most readily available oxidizers have already been described in previous volumes in this series: **Deadly Brew: Advanced Improvised Explosives** (Vol. I) and **Shock Sensitive Industrial Materials: Advanced Improvised Explosives** (Vol. II). They are listed below.

ACETYL BENZOYL PEROXIDE $\text{C}_6\text{H}_5\text{CO-O}_2\text{-OCCH}_3$	VOL II
ACETYL PEROXIDE $(\text{CH}_3\text{CO})_2\text{O}_2$	VOL II
ALUMINUM CHLORATE $\text{Al}(\text{ClO}_3)_3$	VOL II
AMMONIUM NITRITE $\text{NH}_4\text{NO}_2$	VOL II
BARIUM CHLORATE $\text{Ba}(\text{ClO}_3)_2 \cdot \text{H}_2\text{O}$	VOL II

BENZOYL PEROXIDE	VOL I
$C_{14}H_{10}O_4$	
CADMIUM NITRATE	VOL II
$Cd(NO_3)_2$	
CALCIUM NITRATE	VOL II
$Ca(NO_3)_2$	
CERIUM NITRATE	VOL II
$Ce(NO_3)_3$	
CHLORINE HEPTAOXIDE	VOL II
$Cl_2O_7$	
p-CHLOROBENZOYL PEROXIDE	VOL II
$(ClC_6H_4CO)_2O_2$	
CHROMIUM NITRATE	VOL II
$Cr(NO_3)_3$	
FLUORINE NITRATE	VOL II
$FNO_3$	
HYDROGEN PEROXIDE	VOL I
$H_2O_2$	
LAUROYL PEROXIDE	VOL II
$C_{24}H_{46}O_4$	
LITHIUM CHLORATE	VOL II
$LiClO_3$	

LITHIUM NITRATE	VOL II
$\text{LiNO}_3$	
NITRIC ACID	VOL I
$\text{HNO}_3$	
OZONE	VOL II
$\text{O}_3$	
PERCHLORIC ACID	VOL II
$\text{HClO}_4$	
POTASSIUM NITRATE	VOL II
$\text{KNO}_3$	
POTASSIUM NITRITE	VOL II
$\text{KNO}_2$	
SILVER PERMANGANATE	VOL I
$\text{AgMnO}_4$	
SODIUM CHLORITE	VOL II
$\text{NaClO}_2$	
SODIUM NITRATE	VOL II
$\text{NaNO}_3$	
STRONTIUM NITRATE	VOL II
$\text{Sr}(\text{NO}_3)_2$	
STRONTIUM PEROXIDE	VOL II
$\text{SrO}_2$	

SULFURIC ACID	VOL I
$\text{H}_2\text{SO}_4$	
THALLIUM NITRATE	VOL II
$\text{TINO}_3$	
THORIUM NITRATE	VOL II
$\text{Th}(\text{NO}_3)_2$	
ZINC NITRATE	VOL II
$\text{Zn}(\text{NO}_3)_2$	
ZIRCONIUM NITRATE	VOL II
$\text{Zr}(\text{NO}_3)_2$	

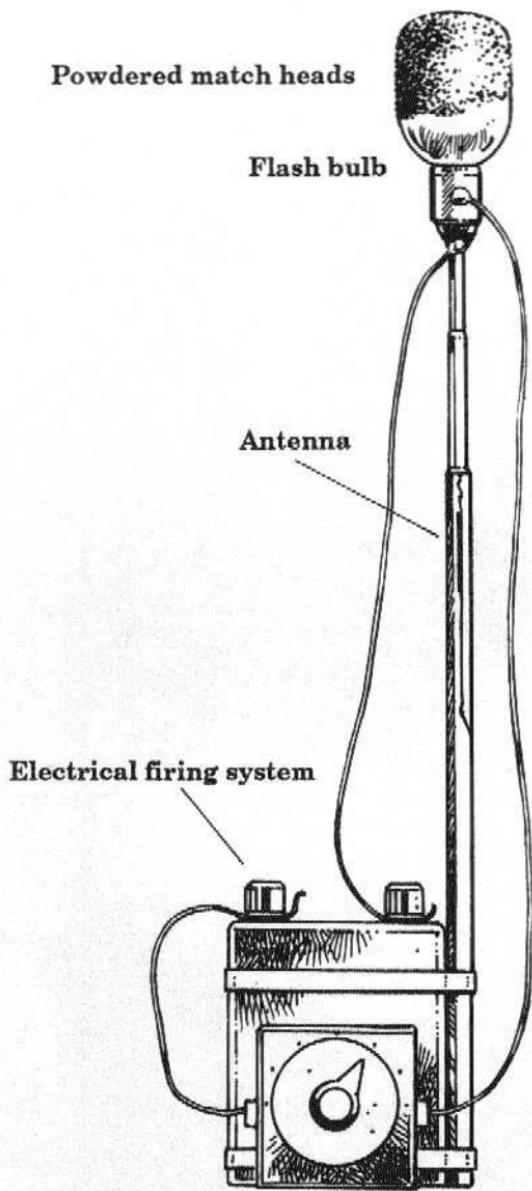
## CHAPTER 3

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# Ignition Devices

### FLAME IGNITER

1. Coat a photographer's flashbulb with epoxy glue and powdered matchheads.
2. Attach a telescoping antenna to the flashbulb.
3. Attach electrical wires to the side and base of the flashbulb. Note: Wires must be long enough to allow the complete extension of the antenna.
4. Attach the antenna to any convenient electrical firing assembly.
5. Wire the flashbulb into the electrical firing assembly.
6. Pour the incendiary liquid on the floor.
7. Place the igniter in the center of the room.
8. Extend the antenna so that the flash will occur well up into the vapor.



**Figure 1. Flame Igniter**

## OXIDIZER IGNITER

1. Fill a bottle with any convenient oxidizing material.
2. Attach an electrical blasting cap and any convenient electrical firing system to the bottle.

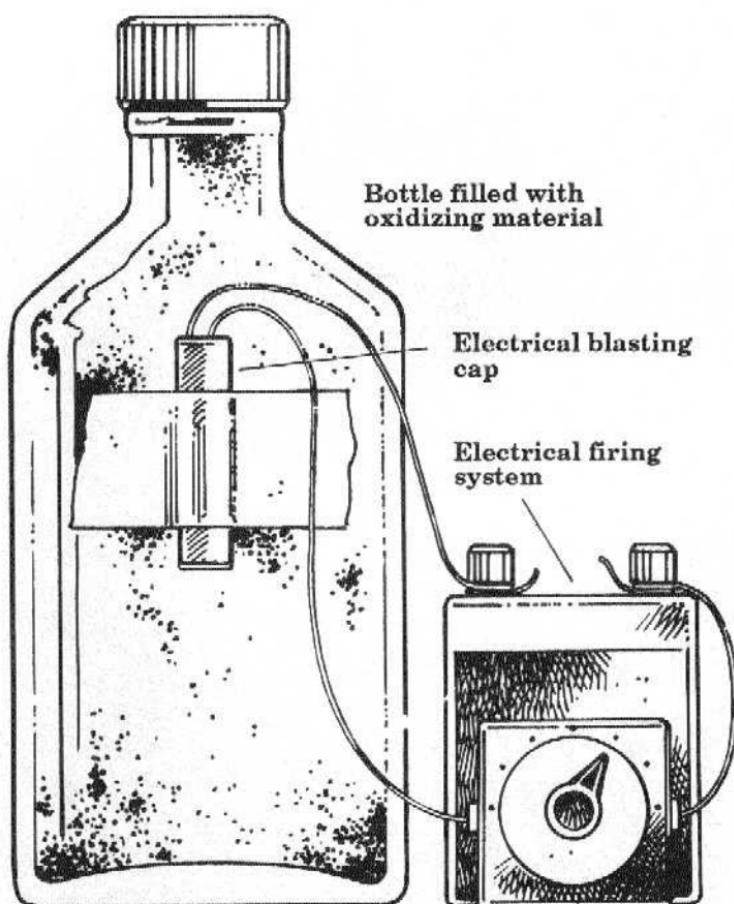


Figure 2. Oxidizer Igniter

3. Pour the incendiary liquid on the floor.
4. Place the igniter in the center of the room.

Make certain that the exterior of the bottle is completely free of oxidizing material.

### **MOLOTOV COCKTAIL - BOOBY TRAP**

1. A mixture of 50% potassium chlorate and 50% powdered sugar is added to warm water to saturation.
2. Soak a clean cotton rag in the mixture, then allow it to dry, leaving the chlorate/sugar absorbed in the rag.
3. Fill a bottle 3/4 full of a mixture comprised of 1/4 sulfuric acid and 3/4 gasoline.
4. Conceal a zinc washer in the cap of the bottle.
5. Put the components together as an ordinary Molotov cocktail. (See Figure 3.)

Leave this device in an obvious location. When the person who finds it attempts to move it, the liquid in the bottle will come into contact with the zinc in the cap and produce hydrogen gas. The pressure will eventually shatter the bottle. The chlorate/sugar will come into contact with the sulfuric acid and ignition will occur.

The safest way to transport this device is to seal it with an ordinary cap. Remove this cap and install the zinc-treated cap when the bottle is in place.

This device—like all booby traps—has the potential to be more dangerous for its maker than its intended target.

Rag is never stuffed into the bottle's opening. This can cause premature ignition or spill flaming liquid on the thrower.

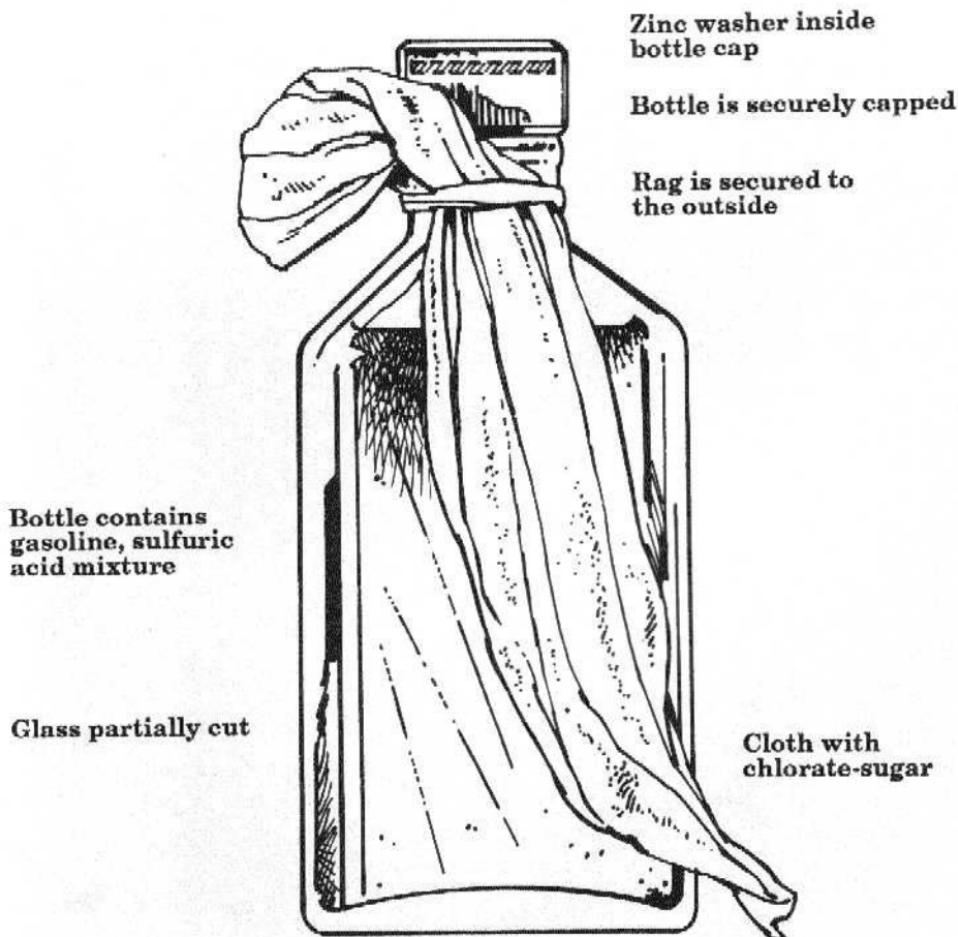


Figure 3. Molotov Cocktail Booby Trap

## CHAPTER 4

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# Labels

To facilitate the safer handling of industrial materials, a system of labels has been devised. The labels are diamond-shaped and divided into four compartments. (See Figure 4.)

The system identifies the hazards of materials in the areas of Health, Flammability, and Reactivity. These categories are rated as follows:

### **HEALTH**

Left Side—Blue

- 4 Materials which on very short exposure could cause death or major residual injury even if prompt medical treatment were given.
- 3 Materials which on short exposure could cause serious temporary or residual injury even if prompt medical treatment were given.
- 2 Materials which on intense or continued exposure could cause temporary incapacitation or possible residual injury unless prompt medical treatment is given.

- 1 Materials which on exposure would cause irritation but only minor residual injury even if no treatment were given.
  - 0 Materials which offer no particular hazard.



**Figure 4. Hazardous Material Label**

#### **FLAMMABILITY**

Top - Red

- 4 Materials which will rapidly or completely vaporize at atmospheric pressure and normal ambient temperature, or which are readily dispersed in air and which will burn readily.
  - 3 Liquids and solids that can be ignited under almost all ambient temperature conditions.
  - 2 Materials that must be moderately heated or exposed to relatively high ambient temperature before ignition can occur.
  - 1 Materials that must be preheated before ignition can occur.
  - 0 Materials that will not burn.

## **REACTIVITY**

Right Side—Yellow

- 4 Materials which in themselves are readily capable of detonation or of explosive decomposition or reaction at normal temperatures and pressures.
- 3 Materials which in themselves are readily capable of detonation or explosive reaction but require a strong initiating source or which must be heated under confinement before initiation or which react explosively with water.
- 2 Materials which in themselves are normally unstable and readily undergo violent chemical change but do not detonate. Also materials which may react violently with water or which may form potentially explosive mixtures with water.
- 1 Materials which in themselves are normally stable, but which can become unstable at elevated temperatures and pressures or which may react with water with some release of energy but not violently.
- 0 Materials which are normally stable.

The fourth (lower) part of the label is used to indicate unusual reactivity with water. This reactivity is indicated by the letter "W" with a line through it. This space may also be used to indicate radioactivity.

## APPENDIX A

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# Recommended Reading

*USMC Destruction by Demolition, Incendiaries and Sabotage*  
Paladin Press

*Improvised Munitions Black Book, Vols. 1 and 2*  
Paladin Press

*Fire Protection Guide on Hazardous Materials*  
National Fire Protection Association

*Hazardous Chemicals Data*  
National Fire Protection Association

*The Science of Industrial Explosives*  
M.A. Cook  
Reinhold Publishing

*Industrial Explosives in America*  
J. Dannenberg  
ABA Publishing

*Condensed Chemical Dictionary*  
G.G. Hawley  
Van Nostrand Reinhold

*Deadly Brew: Advanced Improvised Explosives*  
Seymour Lecker  
Paladin Press

*Shock Sensitive Industrial Materials: Advanced Improvised Explosives.*  
Seymour Lecker  
Paladin Press

*Fire and Explosion Risks*  
E. Von Schwarts  
Griffen and Company

**TM 31-201-1**  
Department of the Army Technical Manual  
Unconventional Warfare Devices and Techniques –  
Incendiaries

**TM 3-336**  
Department of the Army Technical Manual  
Flame Fuels

**FM 3-21**  
Department of the Army Field Manual  
Chemical Accident Contamination Control

## APPENDIX B

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# Glossary

### *Accelerator*

A compound that reduces the time for vulcanization of rubbers or increases the activity of photographic developer.

### *Alkali*

Any substance which in water solution is caustic to the skin.

### *Antioxidant*

An organic compound added to rubber, natural fats and oils, food products, gasoline, and lubricating oils to retard oxidation, deterioration, and gum formation.

### *Azeotrope*

A liquid mixture of two or more substances which behaves like a single substance in that its vapor has the same composition as the liquid.

### *Base*

A compound that can neutralize an acid to form salts.

### *Catalyst*

Any substance of which a fractional percentage affects the rate of a chemical reaction without itself being

consumed or undergoing a chemical change.

### ***Deliqescent***

Tends to absorb water vapor and become liquid.

### ***Demulsification***

The process of destroying an unwanted emulsion.

### ***Denature***

To change the molecular structure of globular proteins by bringing a protein solution to its boiling point and exposing it to acids or alkalies or various detergents. The process reduces the solubility of proteins and prevents crystallization.

### ***Electroplating***

The deposition of a thin layer of metal on an object by passing an electric current through an aqueous solution of salt containing ions of the element being deposited.

### ***Emulsion***

Suspension of an oil or resin in an aqueous liquid, or of an aqueous liquid in an oil.

### ***Epoxy***

An organic compound. A resin.

### ***Ester***

An organic compound corresponding in structure to a salt in inorganic chemistry.

### ***Ether***

A class of organic compounds containing an oxygen atom between two carbon atoms.

***Flash point***

The temperature at which a liquid or volatile solid gives off vapor sufficient to form an ignitable mixture with the air near its surface.

***Hydrocarbon***

An organic compound consisting exclusively of the elements carbon and hydrogen.

***Hydrolysis***

A chemical reaction in which water reacts with another substance to form two or more new substances.

***Initiator***

A substance that is similar to a catalyst, except that it is consumed in the reaction.

***Intermediate***

An organic compound. Chemical stepping-stones between a parent substance and the final product.

***Leveling agent***

A substance which aids in the uniform dispersion of a coating such as paint.

***Mordant***

A substance capable of binding a dye to a textile fiber.

***Organic chemistry***

A branch of chemistry which embraces the majority of compounds of carbon.

***Oxidizer***

Any compound that spontaneously evolves oxygen at or near room temperature.

***Plasticizer***

An organic compound added to a polymer both to facilitate processing and increase the flexibility and toughness of the final product.

***Polymer***

A macromolecule formed by the chemical union of five or more identical combining units called monomers.

***Reagent***

Any substance used in a reaction for the purpose of detecting, measuring, examining, or analyzing other substances.

***Reducing agent***

The opposite of an oxidizer; removes oxygen from a compound.

***Salt***

The compound formed when the hydrogen of an acid is replaced by a metal or its equivalent. The reaction of an acid and a base yields a salt and water.

***Solvent***

A substance capable of dissolving another substance to form a uniformly dispersed mixture.

***Viscosity***

The internal resistance to flow exhibited by a fluid.

***Vulcanize***

To harden and make rubber more durable by chemical means; usually by combining it with sulfur.